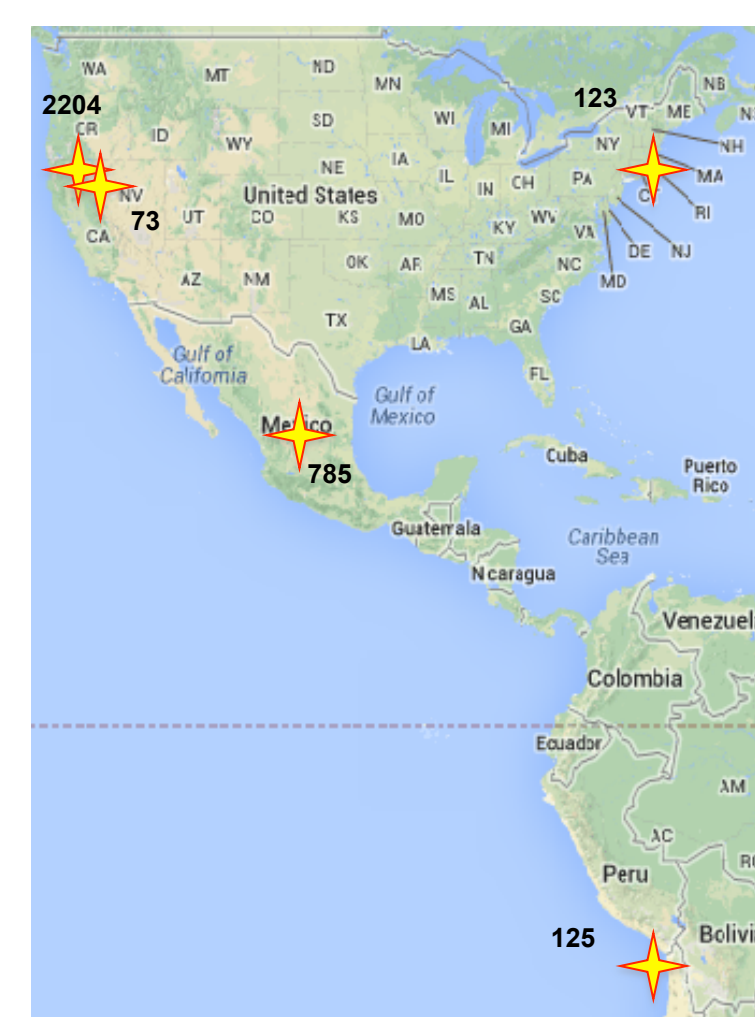


R. O'Brien,^{1,2} Alexander Neu,² Scott A. Epstein,³ Amanda C. MacMillan,³ Steve T. Kelly,² Sergey A. Nizkorodov,³ Alexander Laskin,⁴ R. C. Moffet,¹ M. K. Gilles²

¹ University of the Pacific, ² Lawrence Berkeley National Laboratory, ³ University of California, Irvine, ⁴ Pacific Northwest National Laboratory

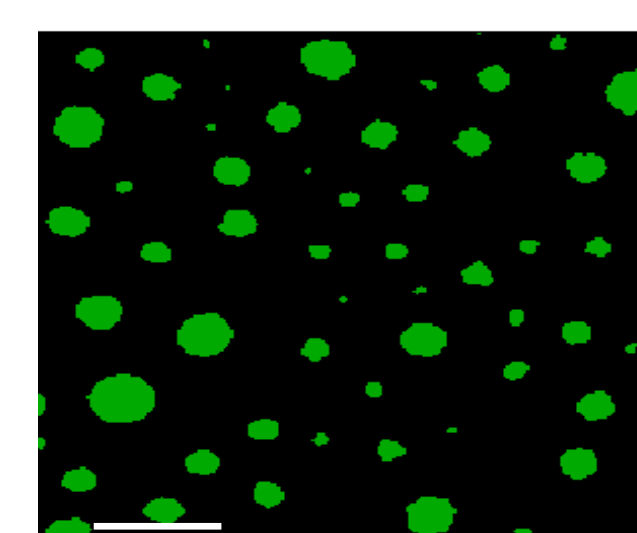
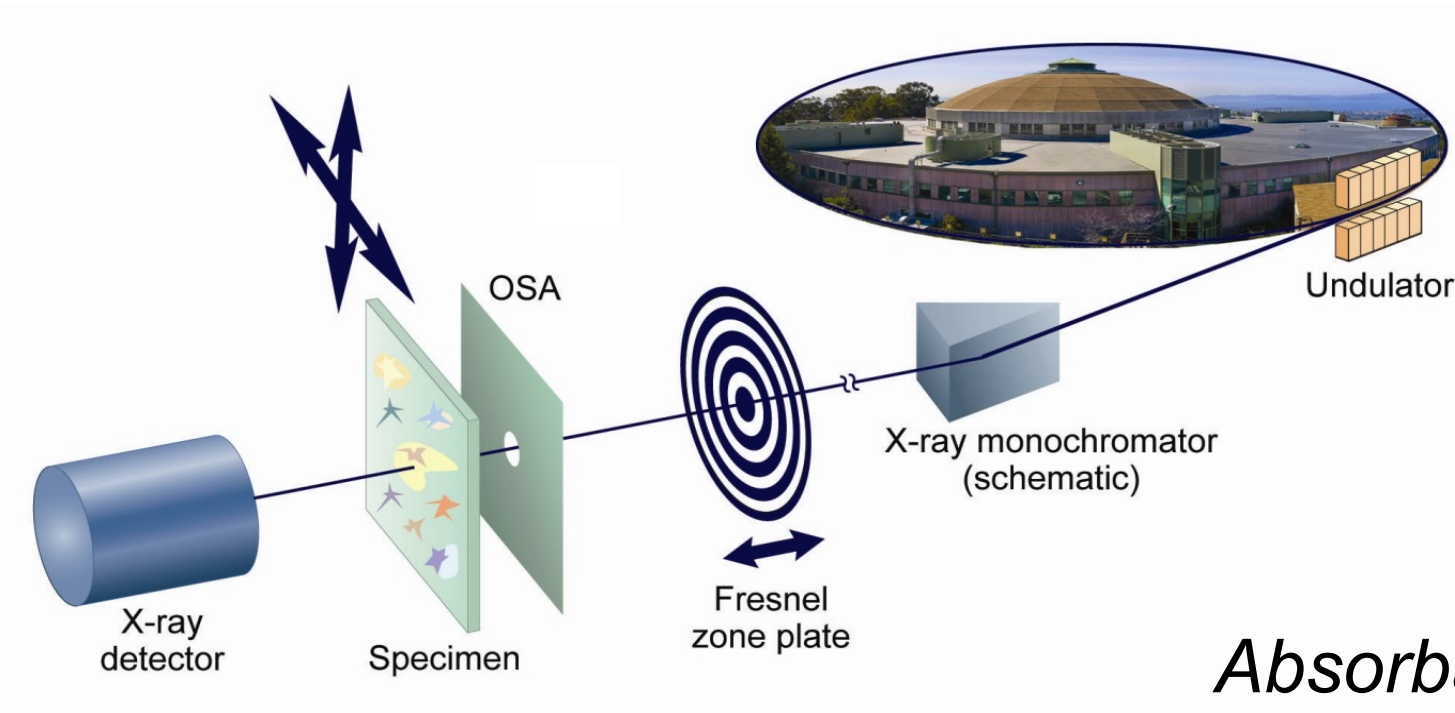
Introduction

- Physical properties of atmospheric aerosols affect their hygroscopicity, ice nucleation ability, the rate of and types of chemical reactions, etc.
- Recent work has shown that organic aerosols can exist as glassy particles under certain atmospheric conditions.¹⁻⁵
- Here, we investigate the viscosity/surface tension of ambient particles from five field campaigns and compare their physical properties to aerosol particles generated in the laboratory under different conditions.



Methods

STXM/NEXAFS



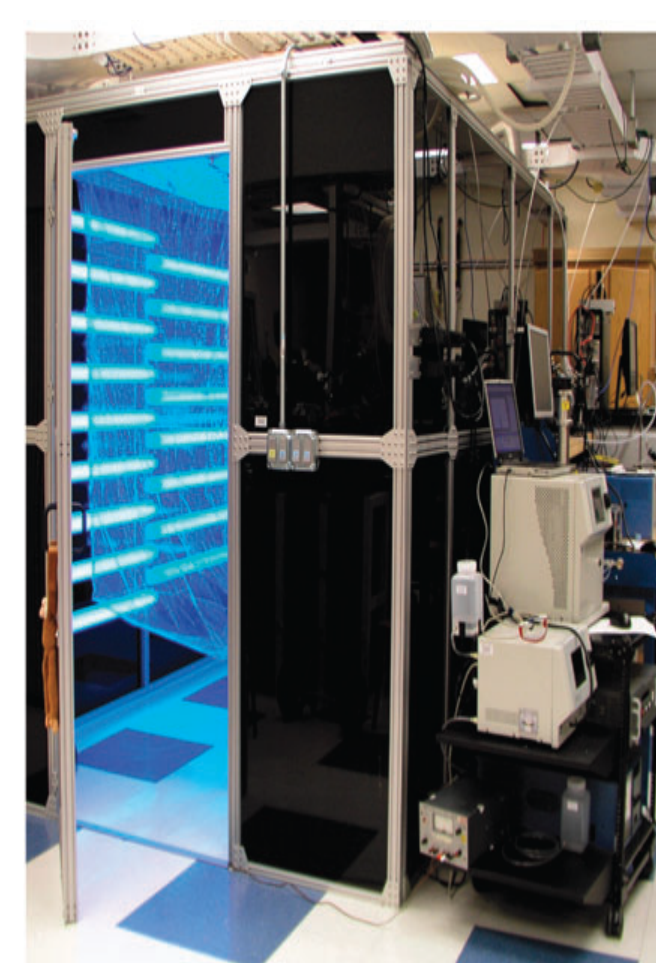
$$\text{Absorbance}(OD) = \ln\left(\frac{I_0}{I}\right) \approx \mu(E) \cdot \rho \cdot d$$

Upon impaction:

Particles with higher viscosity deform less



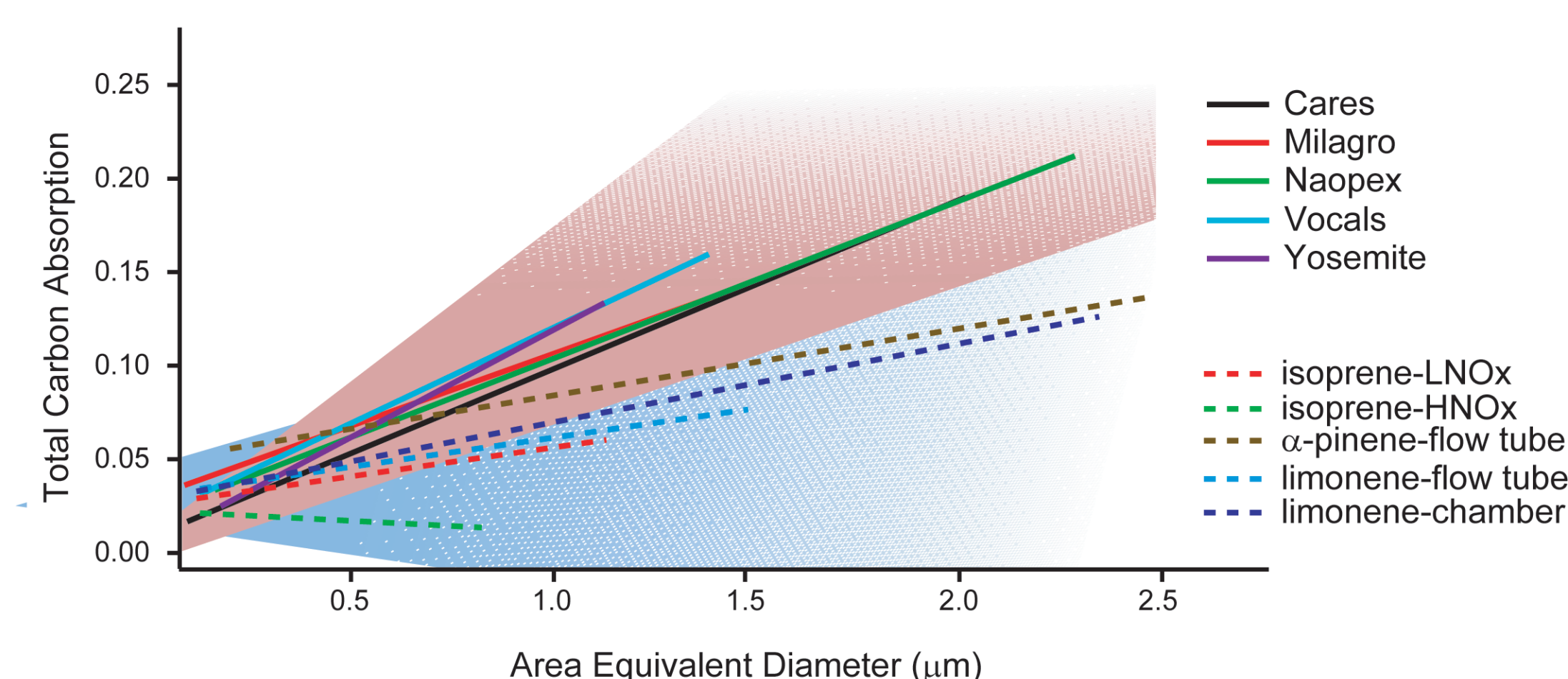
Laboratory generated particles



- Isoprene, α -pinene, d-limonene
- Flow tube and 5 m³ Teflon chamber
- Conditions: high/low NO_x, high/low RH, SO₂, NH₃

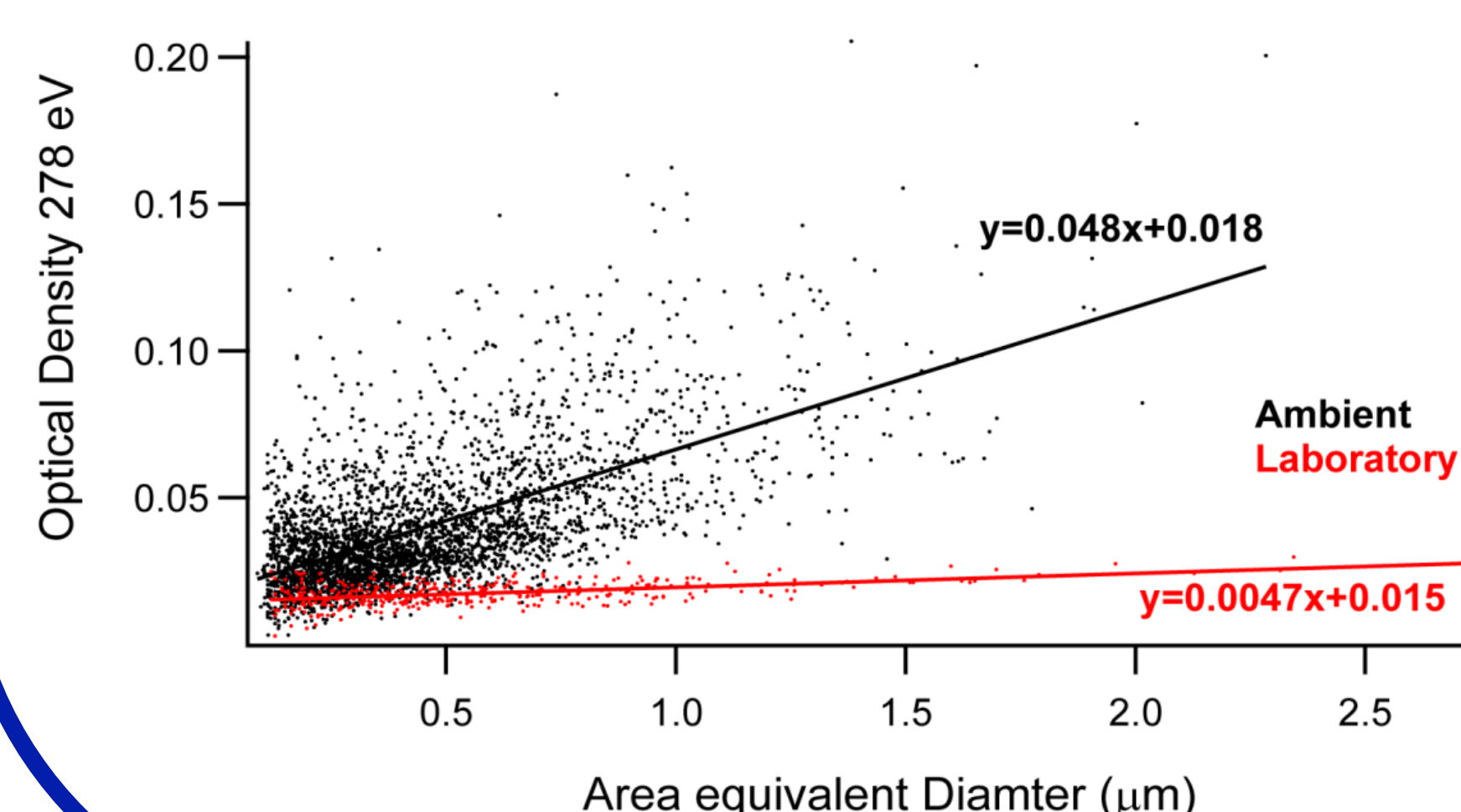
Data/Analysis

Ambient vs. Laboratory



- Optical thickness of carbon (total carbon absorption) as a function of the size of the impacted organic particles.
- Red and blue triangles highlight the \pm 95% confidence intervals for ambient and laboratory generated samples.
- Laboratory generated particles \rightarrow smaller slopes = less viscous/lower surface tension
- High NO_x conditions produced the lowest slope.

Pre-edge analysis



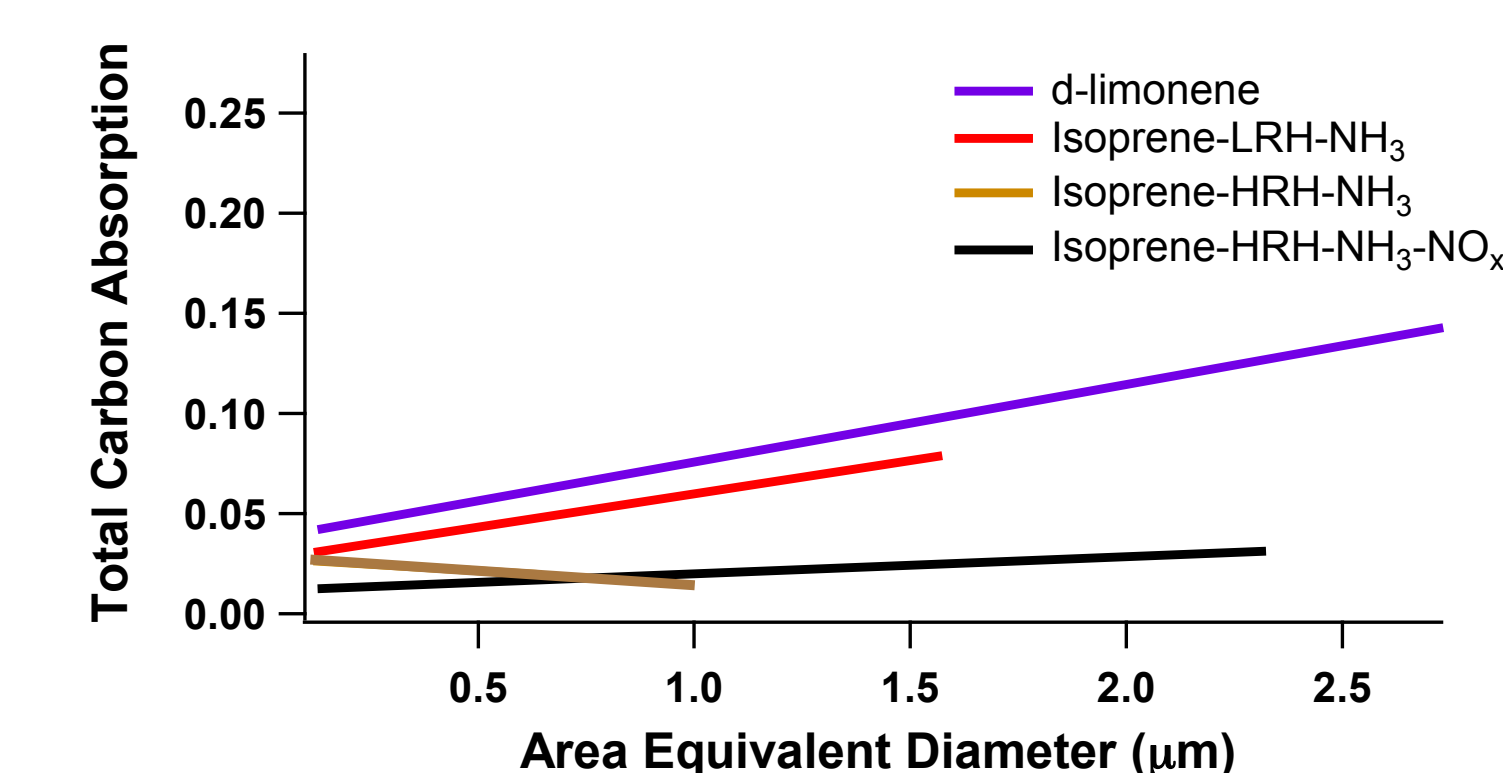
- Ambient particles have higher pre-edge \rightarrow higher inorganic content.
- 11-30% inorganic components in ambient

Acknowledgements

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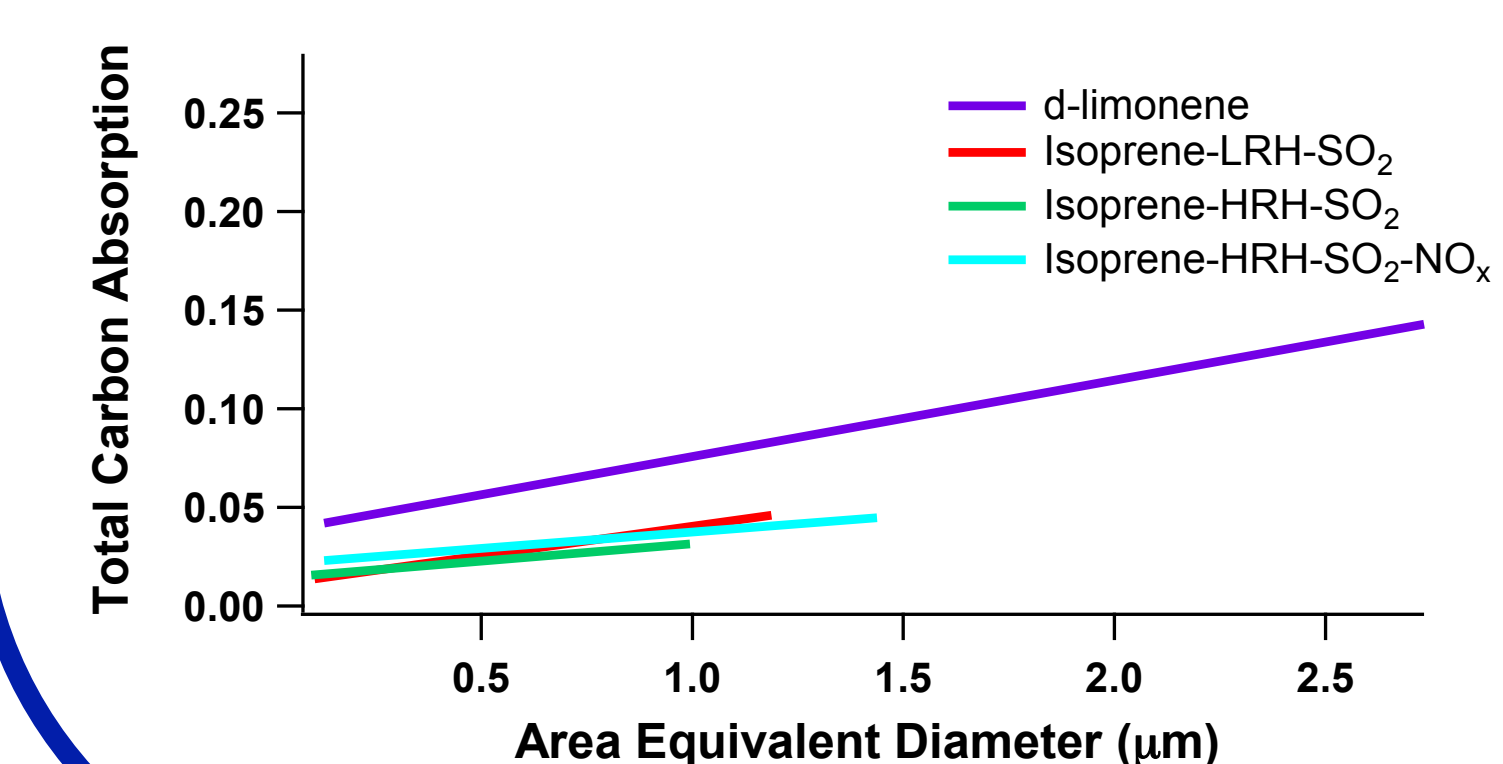
Data/Analysis

Laboratory inter-comparison



- Particles generated with NH₃ in the chamber.
- High NO_x conditions again show decreased slope

- High RH sample \rightarrow negative slope. HRH + NO₃ = lower viscosity



- Particles generated with SO₂ in the chamber.
- Slopes are relatively unchanged \rightarrow SO₂ had minimal impact on viscosity

Conclusions

- STXM/NEXAFS analysis of the size and optical density provides information on the viscosity/surface tension of aerosol particles.
- Organic dominated particles from field campaigns were identified and compared to laboratory generated particles.
- Laboratory generated particles showed lower viscosity than ambient particles.
- Ambient particles have 11-30% inorganic components.
- Neither the addition of NH₃ nor SO₂ to the lab chamber resulted in higher viscosities.

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