# Laboratory Measurements of Aerosol Scavenging in a Cloudy, Turbulent Environment

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## Abstract

How are aerosol particles removed from the atmosphere? Those with diameters greater than a few microns have an appreciable negative vertical velocity and can be removed by settling. Particles smaller than few tens of nanometers have a high diffusivity and can be removed via diffusion. However, the in-between range of a few tenths of a micron has a small settling velocity and a low diffusivity, which produces a bottleneck in removal. As a consequence, this size range is frequently called the "accumulation mode" because particles in it accumulate in the atmosphere. The primary removal mechanism for the accumulation mode is cloud processing, because those particles readily serve as cloud condensation nuclei. The probability that an aerosol particle activates is traditionally understood to depend upon size and chemical composition, described in Köhler theory.

But, turbulent fluctuations in the scalar fields which couple aerosols and cloud droplets (i.e. temperature and water vapor) must also be considered. Measurements from Michigan Tech's turbulent mixing chamber (the Pi Chamber), show that, in the presence of turbulent fluctuations, the correspondence between size and activation is not clearly defined. In steady-state, turbulent cloud conditions a comparison of distribution of interstitial aerosol to the distribution of cloud droplet residuals shows that some aerosol particles are just as likely to remain as interstitial as they are to be activated, a result of fluctuations in the saturation ratio.







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