

Field measurements of the impact of the reactive uptake of reduced nitrogen on the hygroscopicity and light absorption of ambient secondary organic aerosol



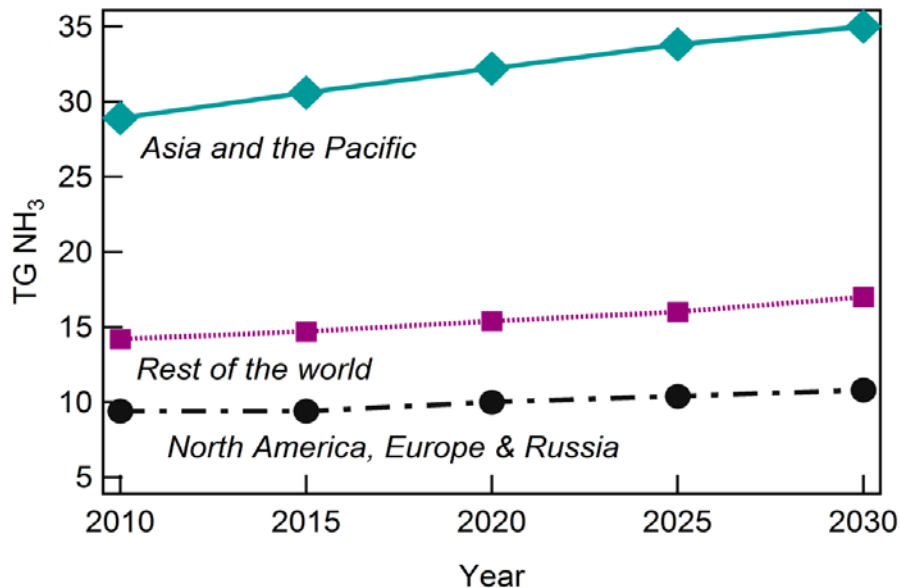
Katy Zimmermann, Xiaolu Zhang

Lynn Russell, Jason Surratt, Karena McKinney, Chris Cappa, Tim Bertram



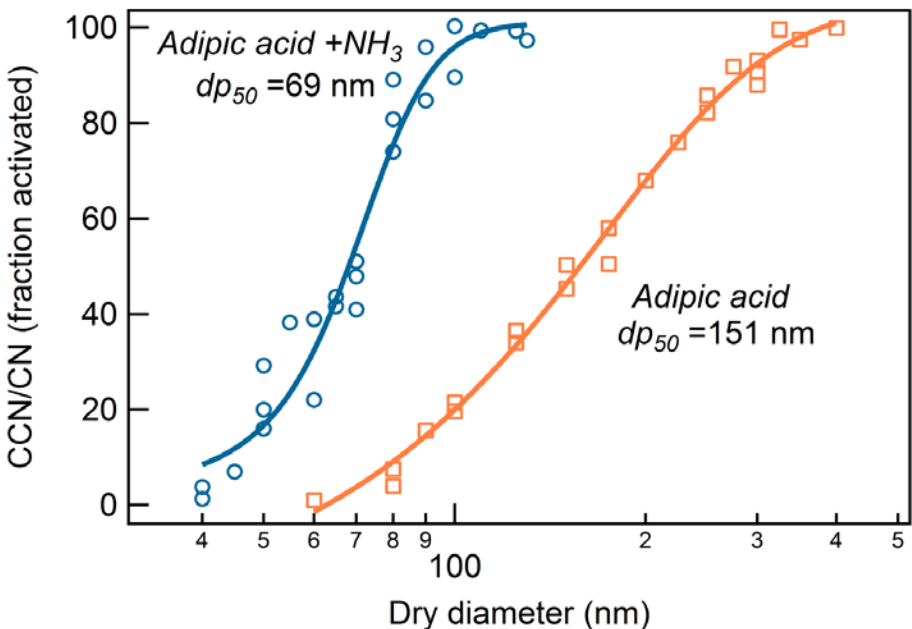
Department of Chemistry
UNIVERSITY OF WISCONSIN-MADISON

Trends in NH₃ Emissions & Potential Impacts



Ammonia emissions are estimated to increase as use of synthetic fertilizers becomes more popular and per capita meat and dairy consumption increase among developing countries.

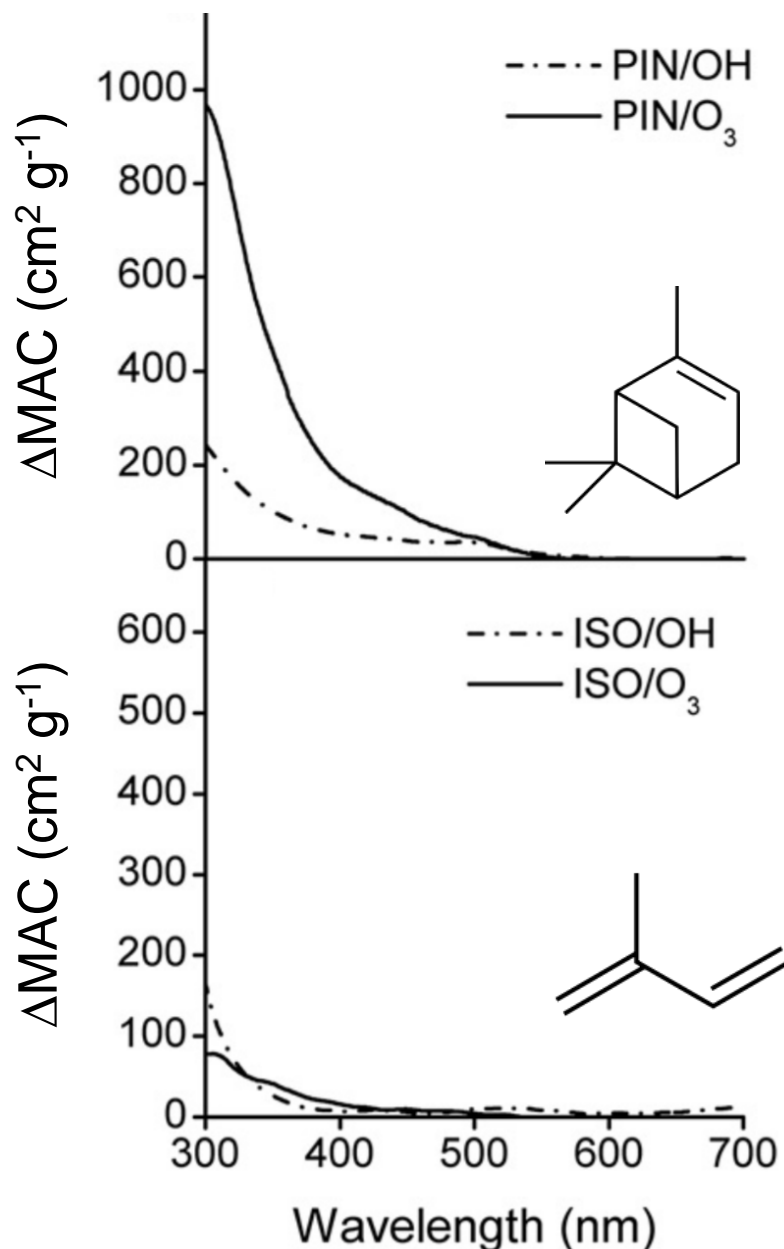
Amann et al., A. Rev. Env. Res. 2013



Multi-phase reactions of NH₃/NH₄⁺ have been shown to **enhance** the hygroscopicity of laboratory generated aerosols composed of slightly soluble organic acids *via* the formation of organic acid-derived ammonium salts.

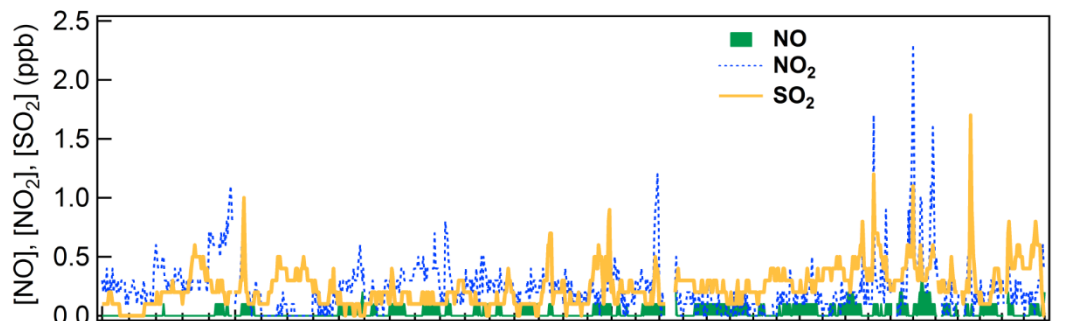
Dinar et al., ES&T 2008

Trends in NH₃ Emissions & Potential Impacts

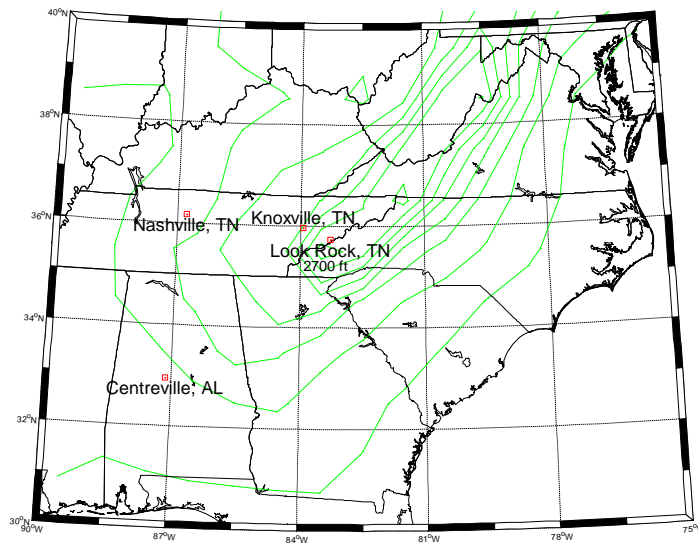
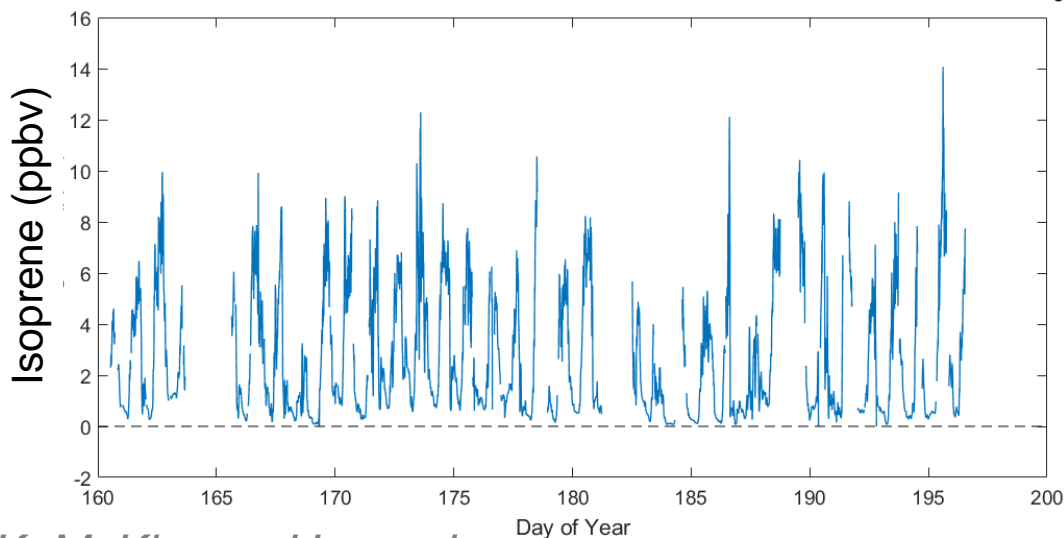
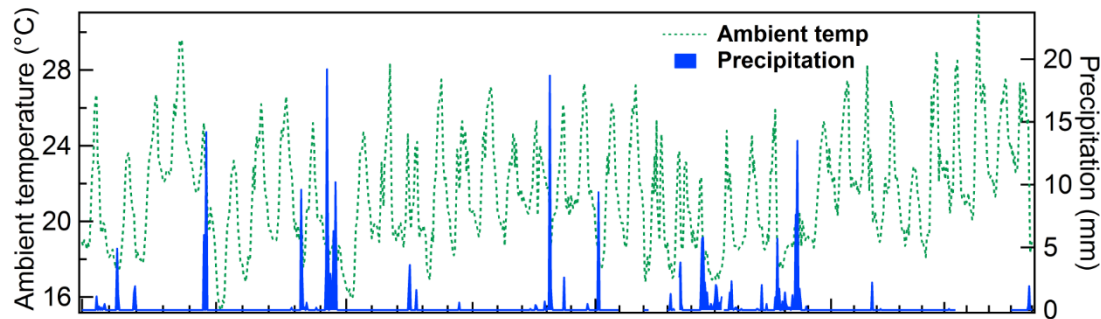


Multi-phase reactions of NH₃/NH₄⁺ have been shown to **increase** the mass absorption coefficient (MAC) for secondary organic aerosol *via* the formation of conjugated nitrogen containing species.

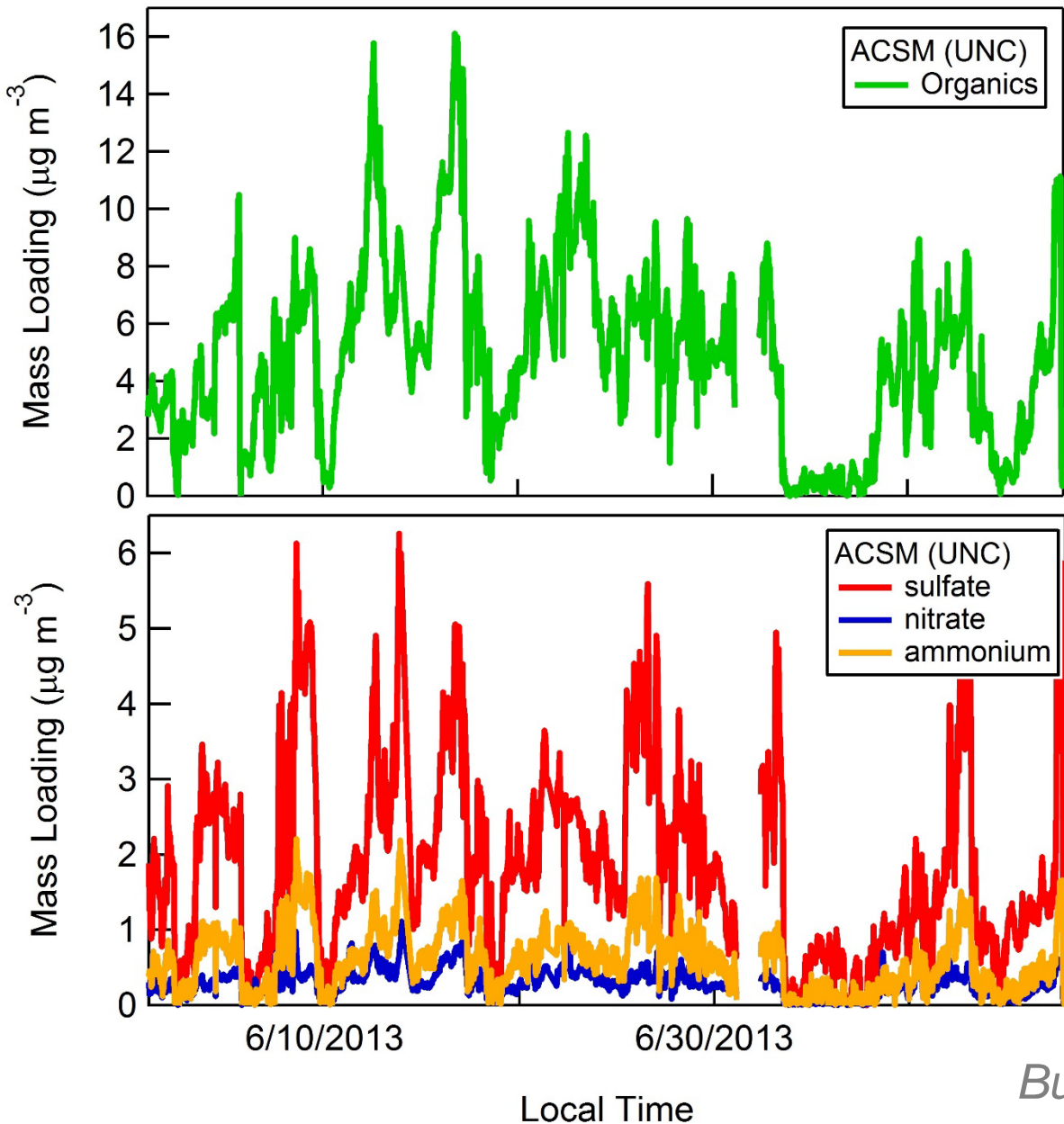
June in Look Rock, TN



[NO_x] < 250 ppt
[O₃] = 25 – 50 ppb
[Isoprene]_{max} = 4-12 ppbv
T_{max} = 22-28 °C



Particle Chemical Composition



Campaign Averages:

$$\text{PM}_{10} = 7.6 \pm 4.7 \mu\text{g m}^{-3}$$

$$f_{\text{organic}} = 0.64$$

$$f_{\text{sulfate}} = 0.24$$

$$f_{\text{ammonium}} = 0.08$$

$$f_{\text{nitrate}} = 0.04$$

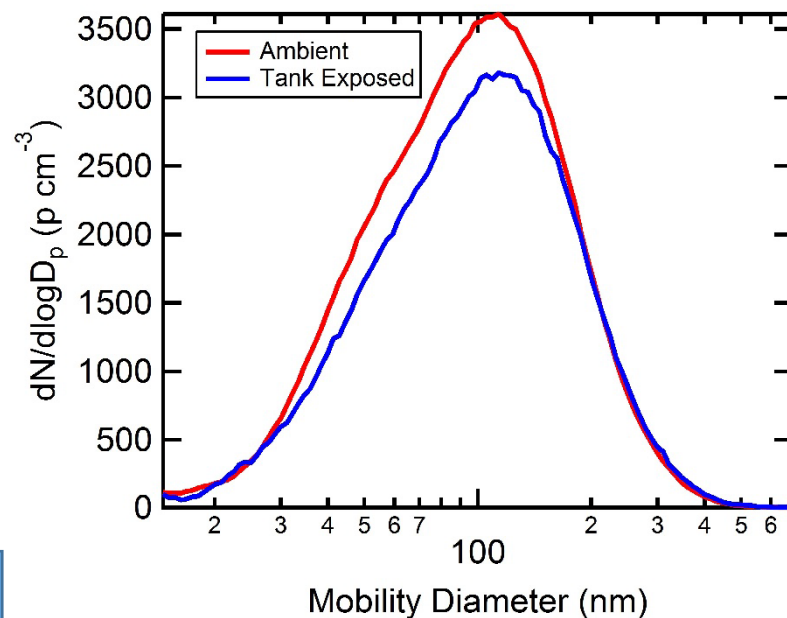
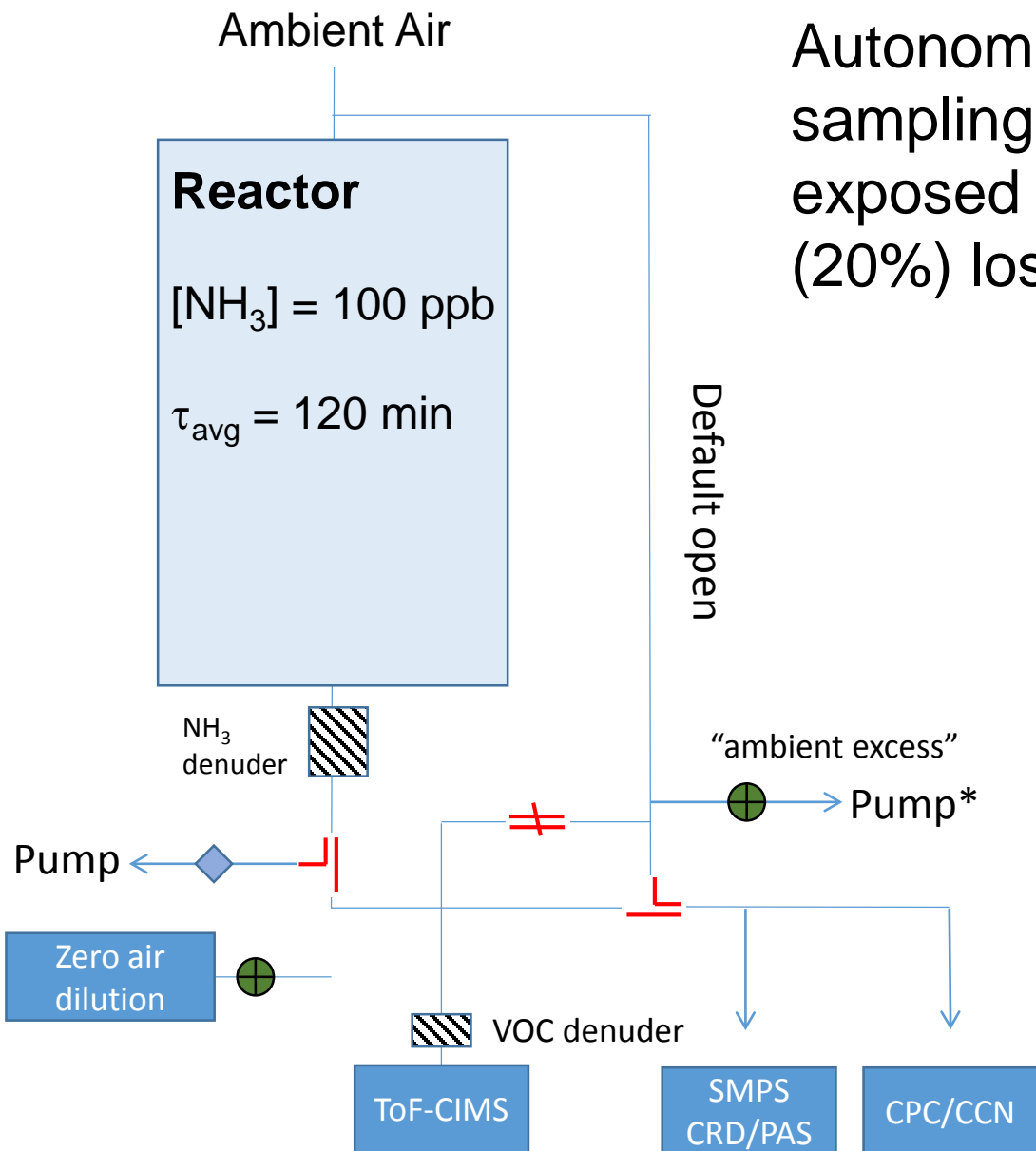
$$\text{O:C} = 0.77 \pm 0.12$$

$$\text{pH} = 1.78 \pm 0.53$$

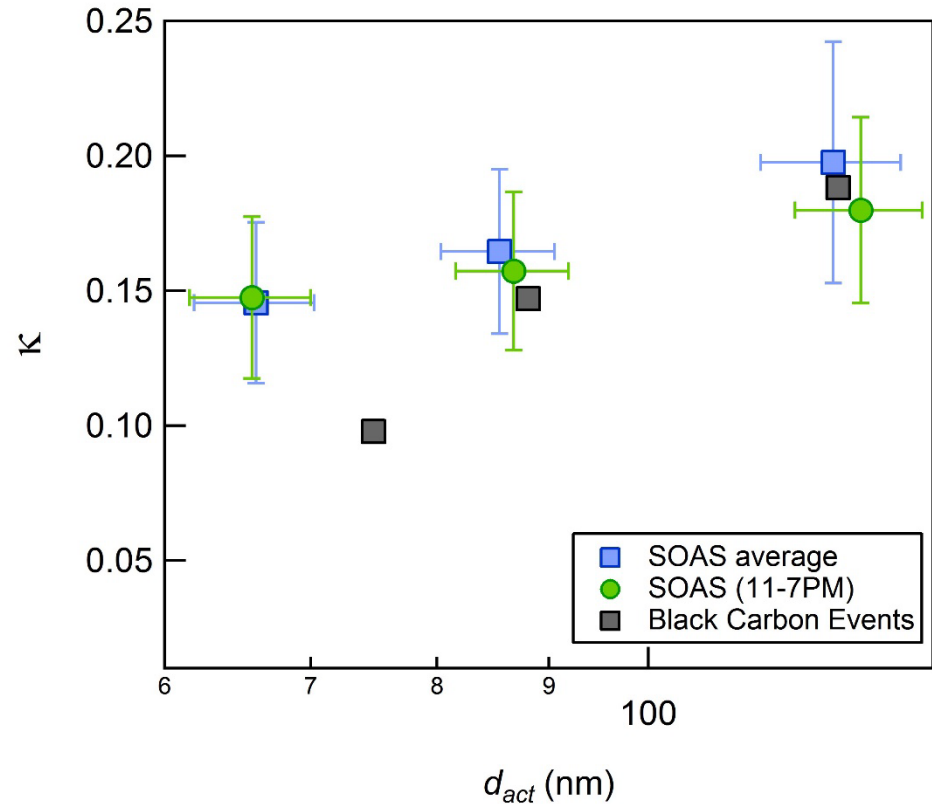
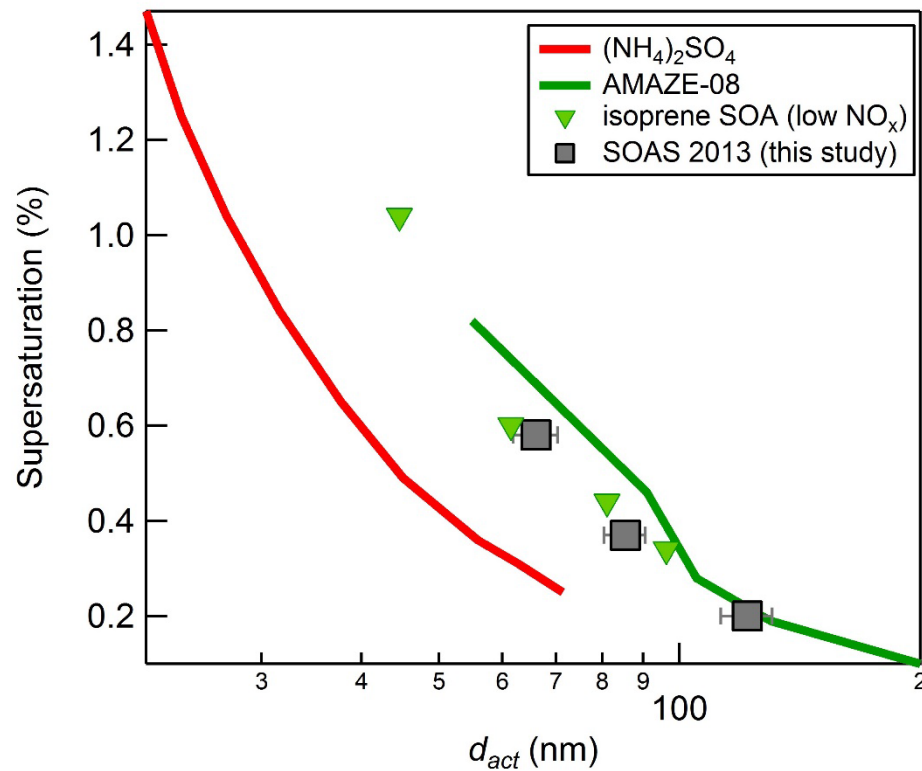
Adapted from
Budisulistiorini et al., ACP 2015

Ammonia Perturbation Experiment

Autonomous duty cycle permits sampling of ambient and NH_3 exposed particles, with measured (20%) loss in particle number.

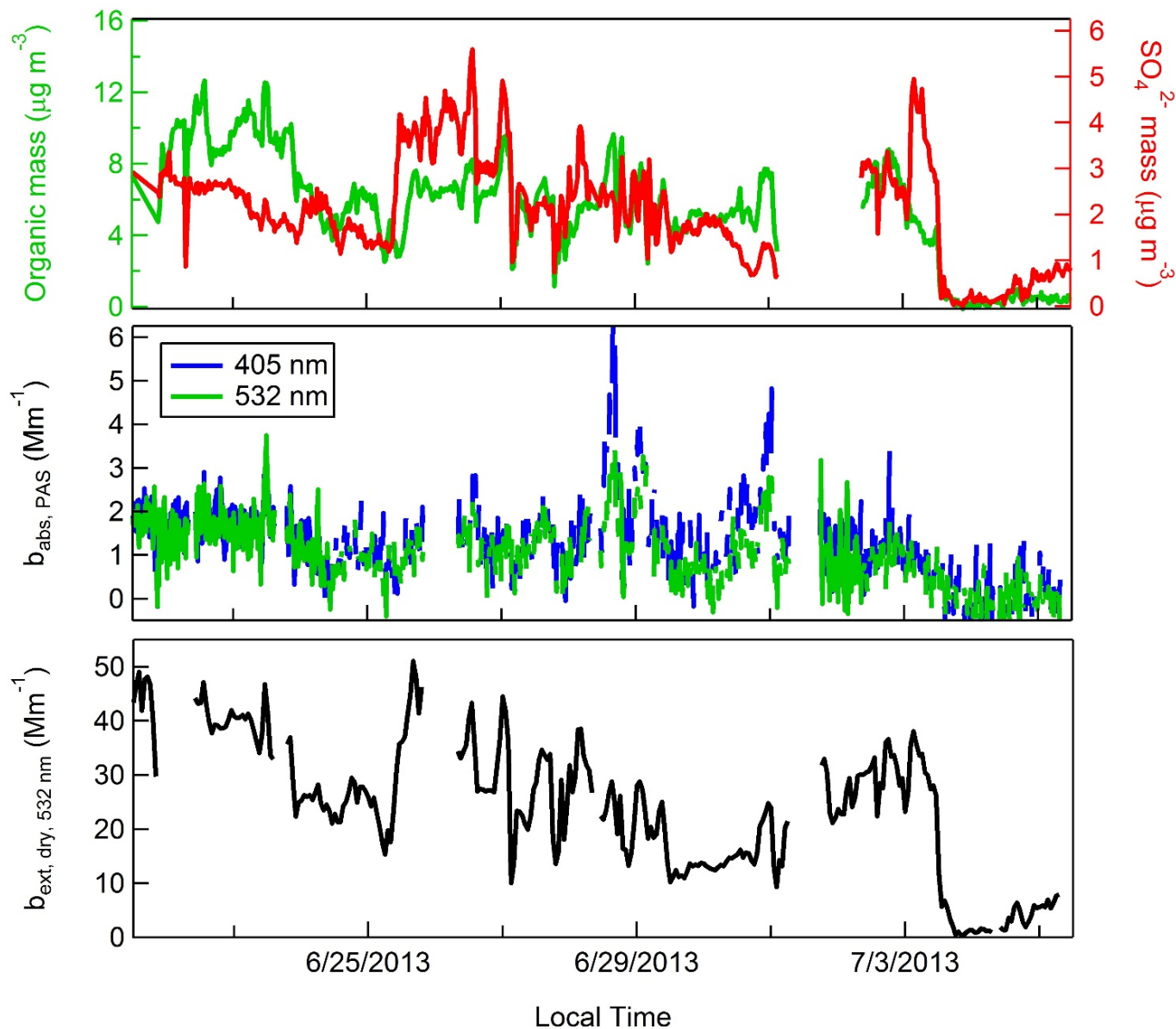


Ambient Particle Hygroscopicity

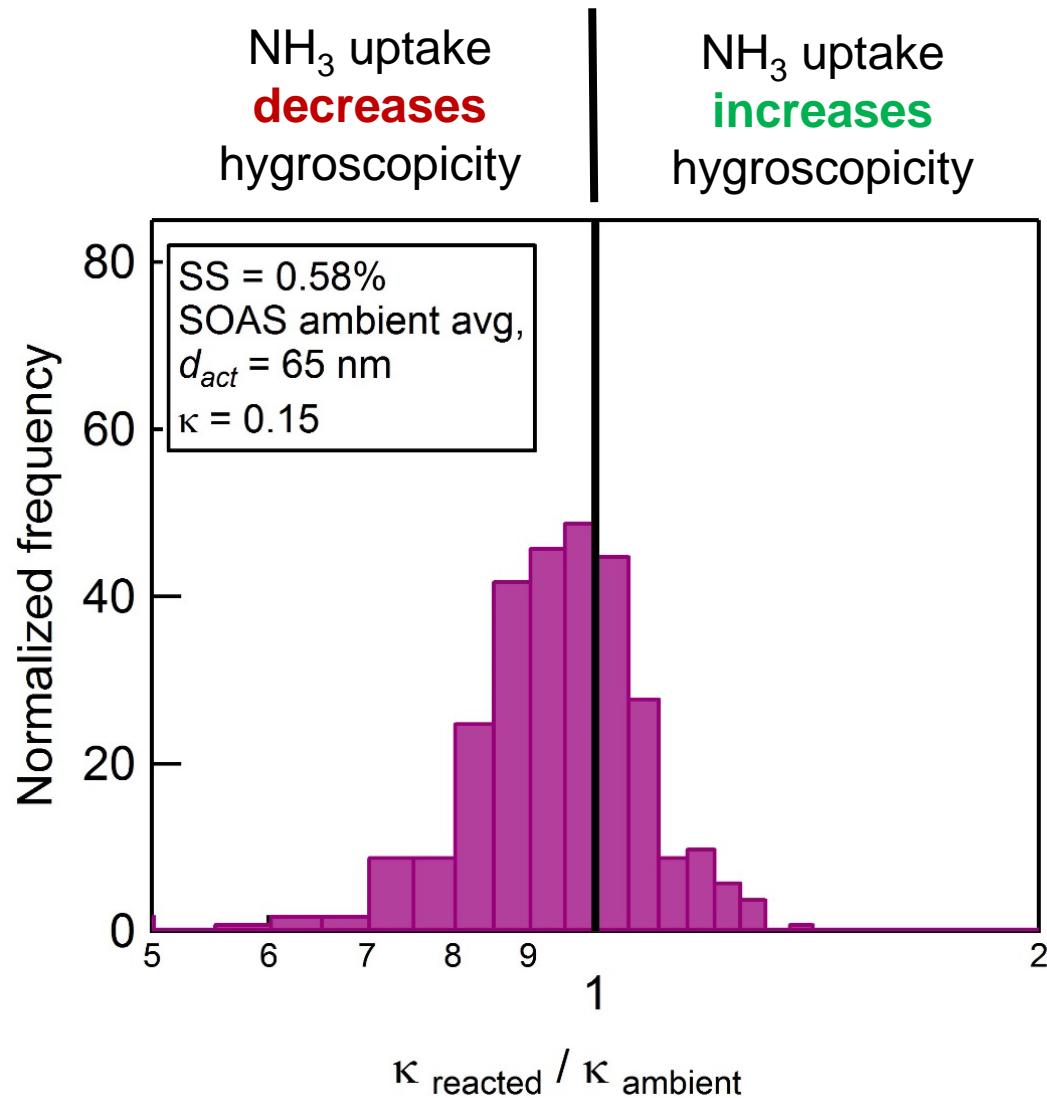


- 1) Ambient determinations of supersaturated hygroscopicity are in agreement with ambient measurements in regions dominated by isoprene chemistry and chamber determinations.
- 2) Size-dependent κ values indicate size-dependent organic mass concentrations of aerosols at Look Rock (*more on poster*).

Ambient Light Extinction and Absorption

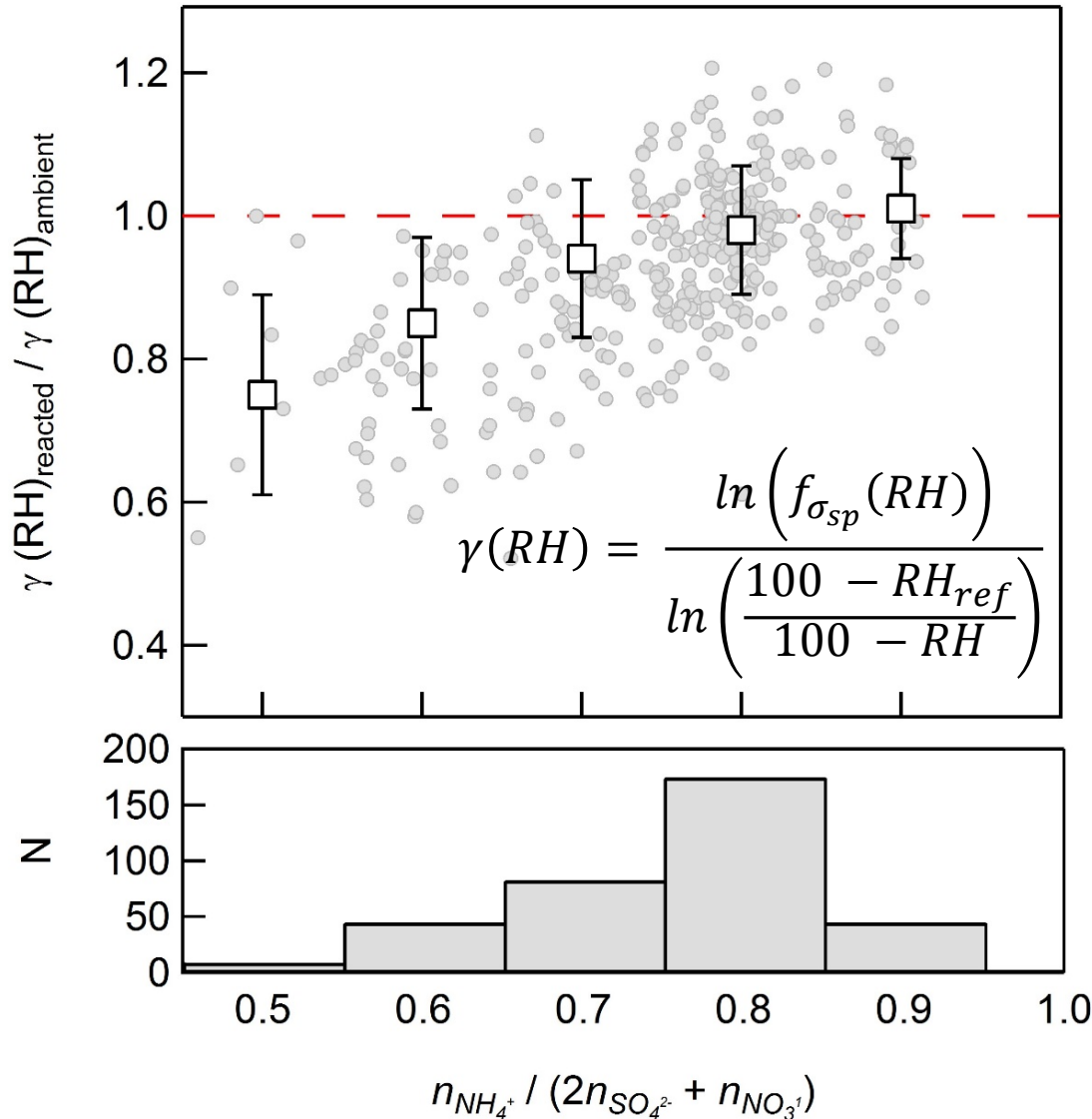


Measured Impact on Hygroscopicity



Particles exposed to NH₃ are **less hygroscopic** than ambient particulates. The net impact on hygroscopicity is small (< 20%).

Measured Impact on Hygroscopicity

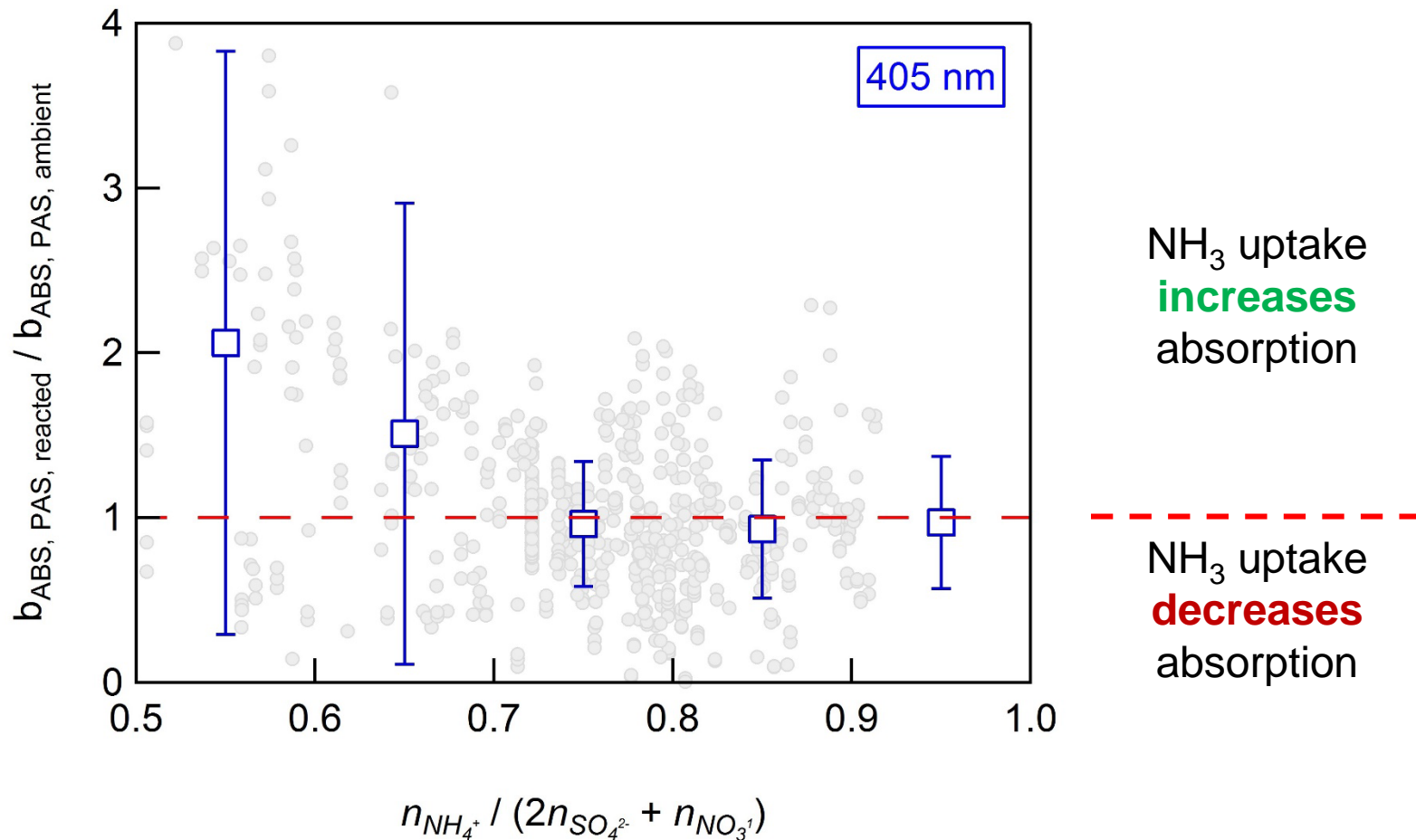


NH₃ uptake
increases
hygroscopicity

NH₃ uptake
decreases
hygroscopicity

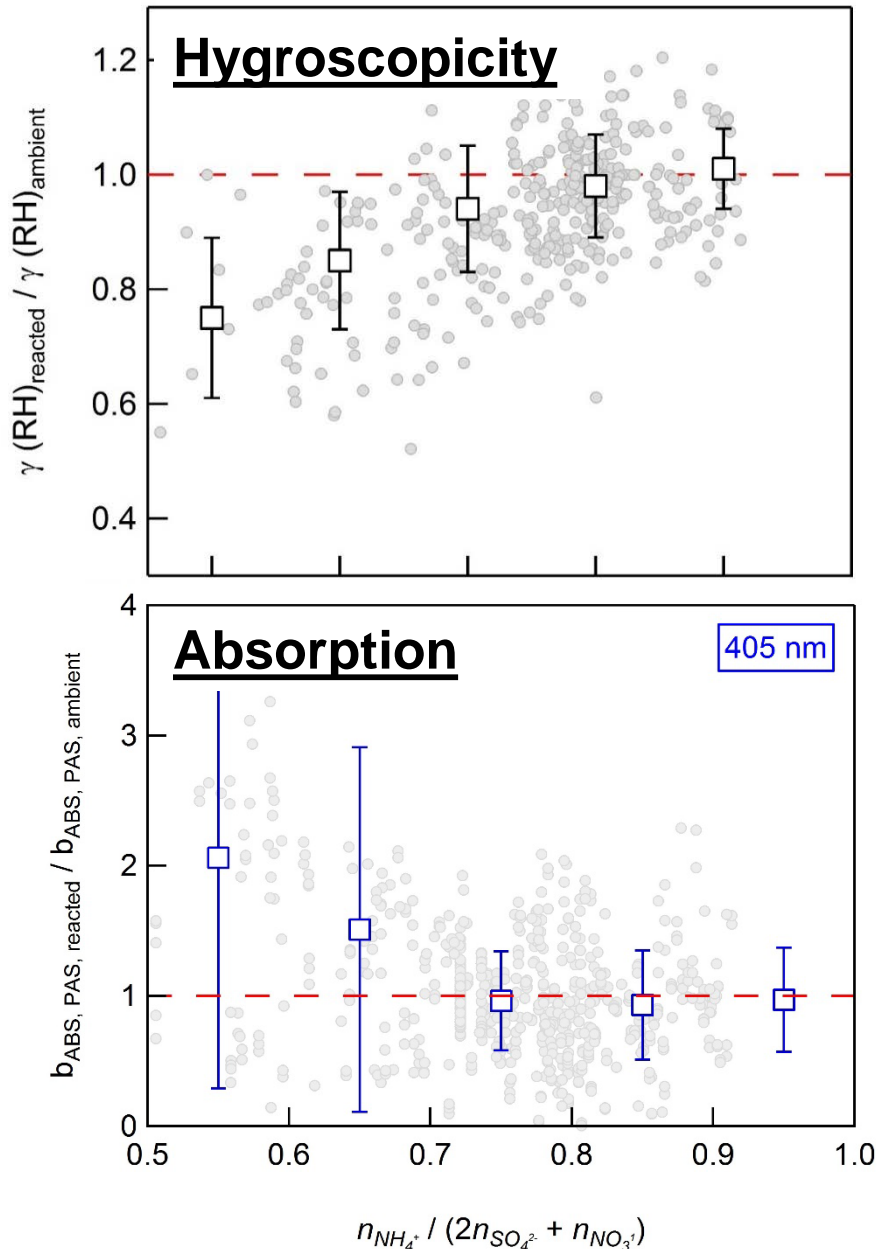
Net NH₃ reactive uptake is observed at low ratios of NH₄⁺ / (SO₄²⁻ + NO₃⁻) leading to a decrease in particle hygroscopicity.

Measured Impact on Light Absorption



Net NH₃ reactive uptake is observed at low ratios of NH₄⁺ / (SO₄²⁻ + NO₃⁻) leading to a increase in absorption at 405nm and has no effect on absorption at 532nm.

Conclusions and Acknowledgements



Conclusions:

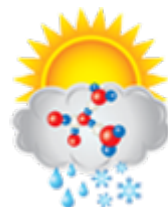
Net NH_3 uptake is observed at low ratios $NH_4^+ / (SO_4^{2-} + NO_3^-)$ leading to a measurable:

Decrease in hygroscopicity

Increase in absorption at 405 nm

Acknowledgements:

Katy Zimmermann, Xiaolu Zhang, Lynn Russell, Jason Surratt, Karena McKinney, Chris Cappa.

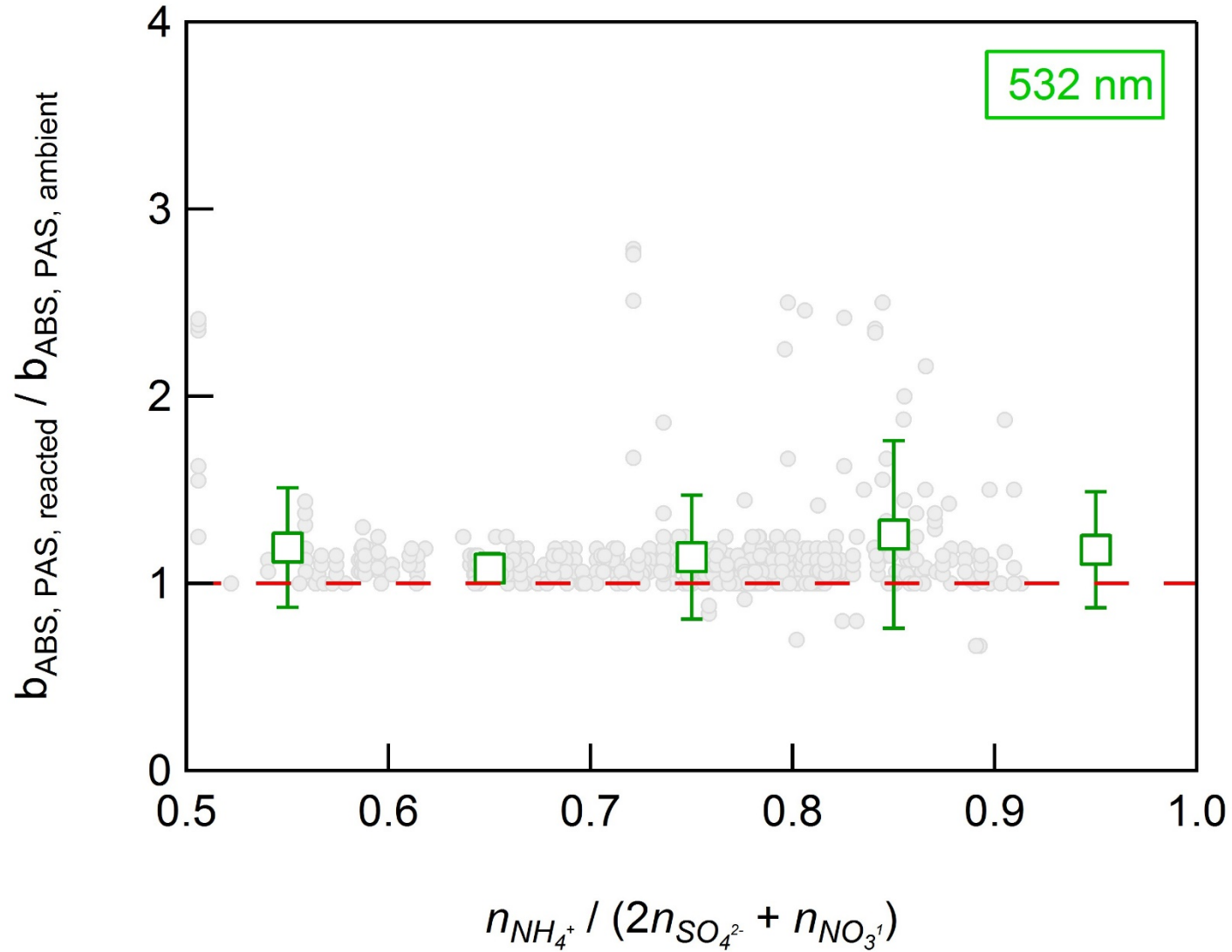


ASR
Atmospheric
System Research

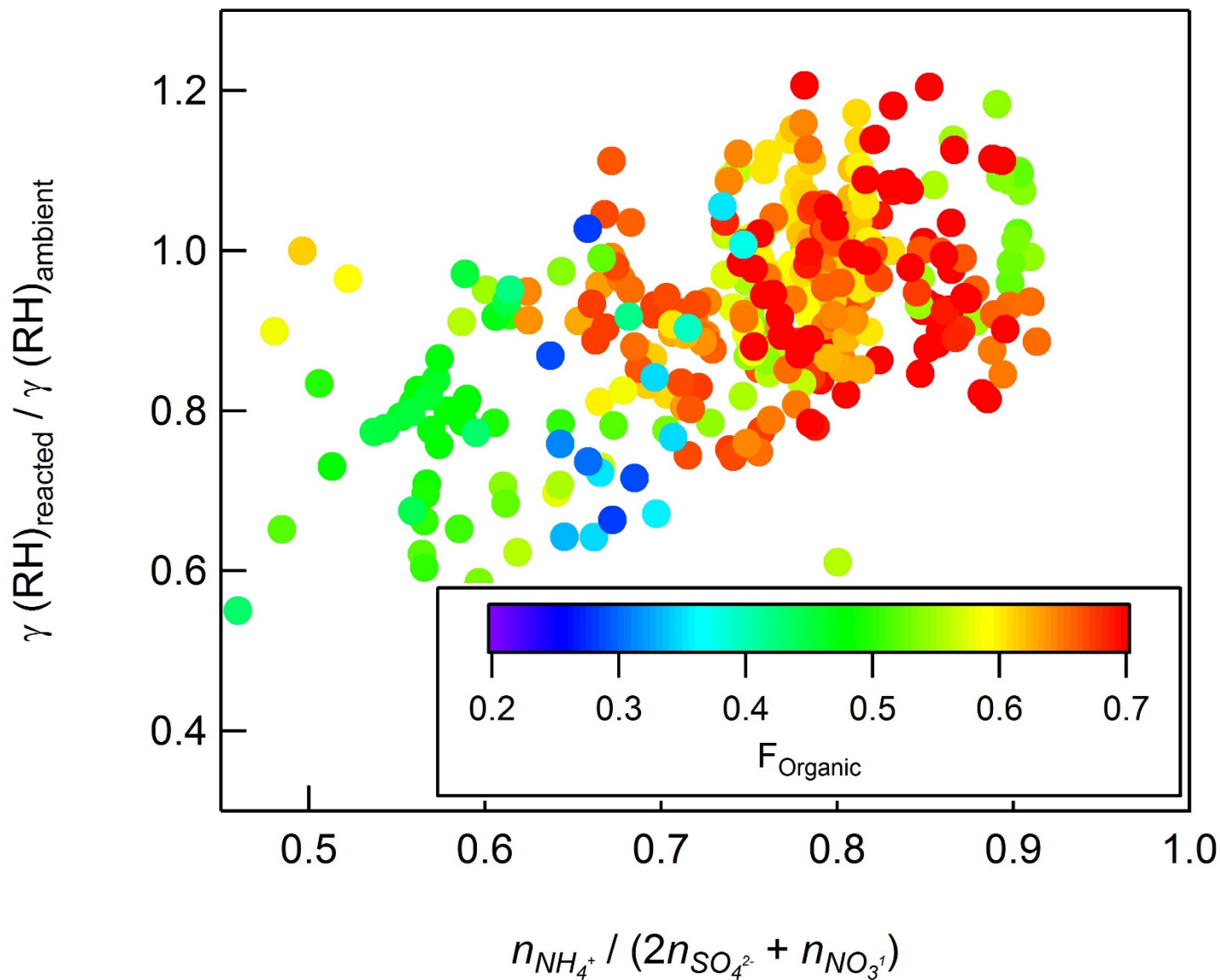


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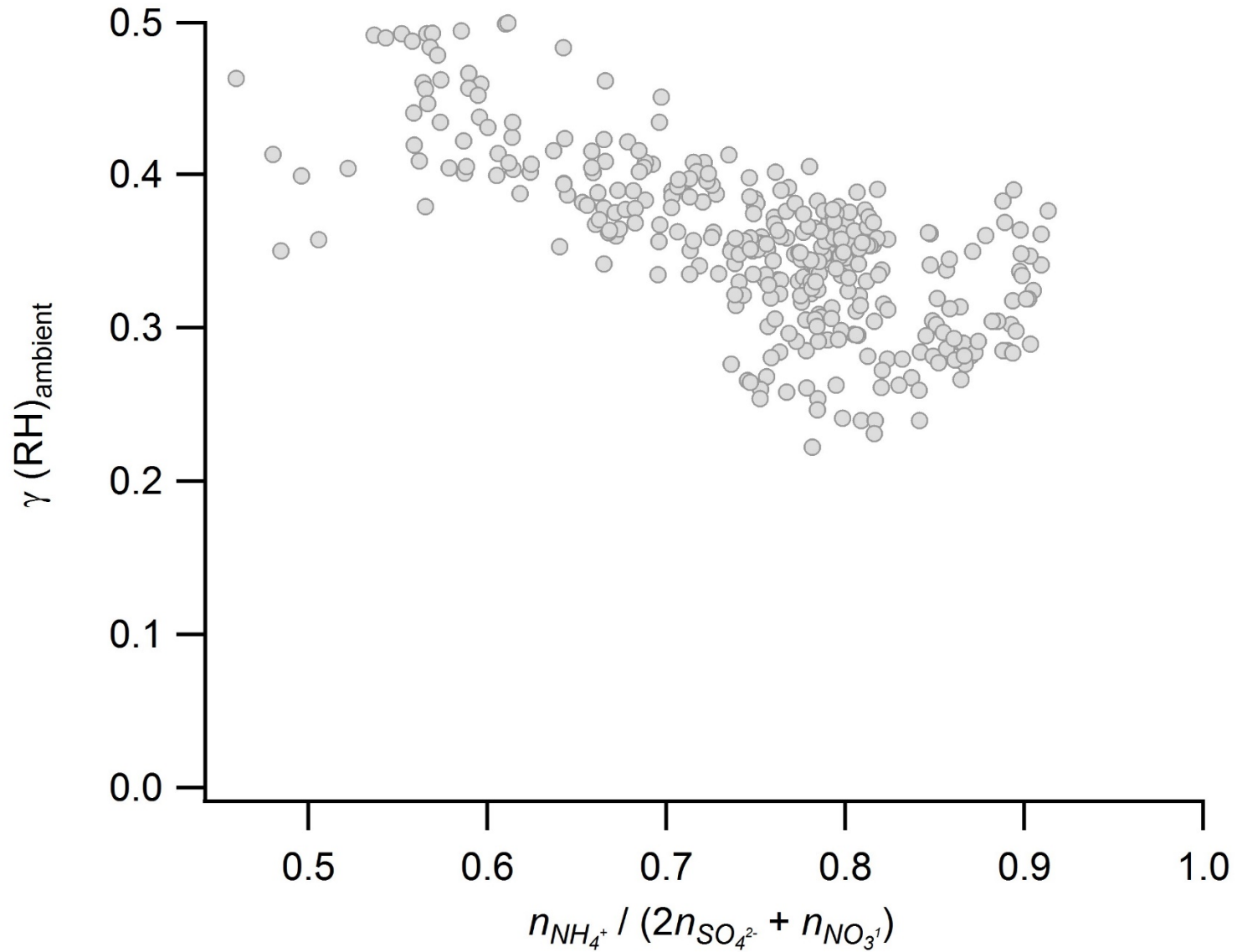
Extra Slides



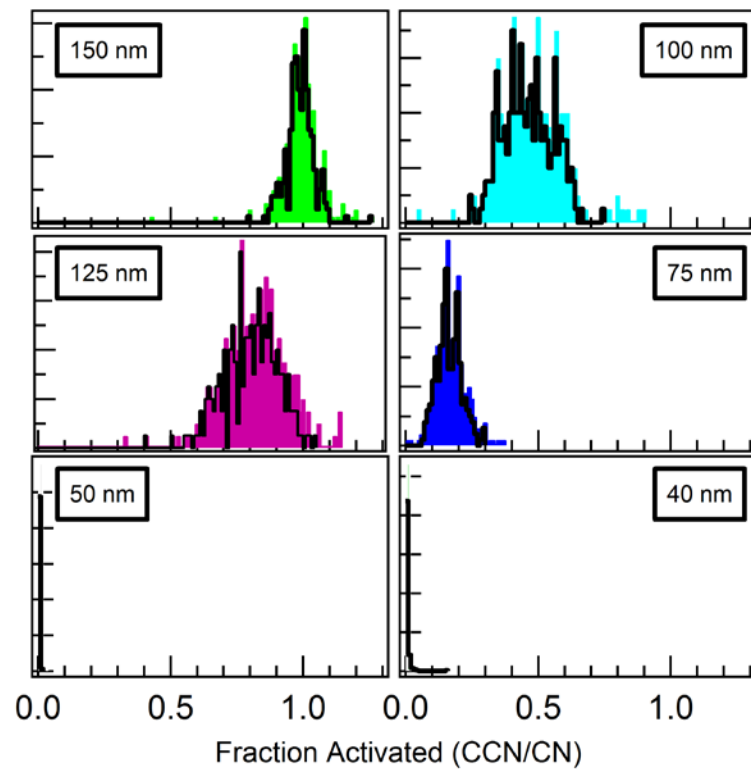
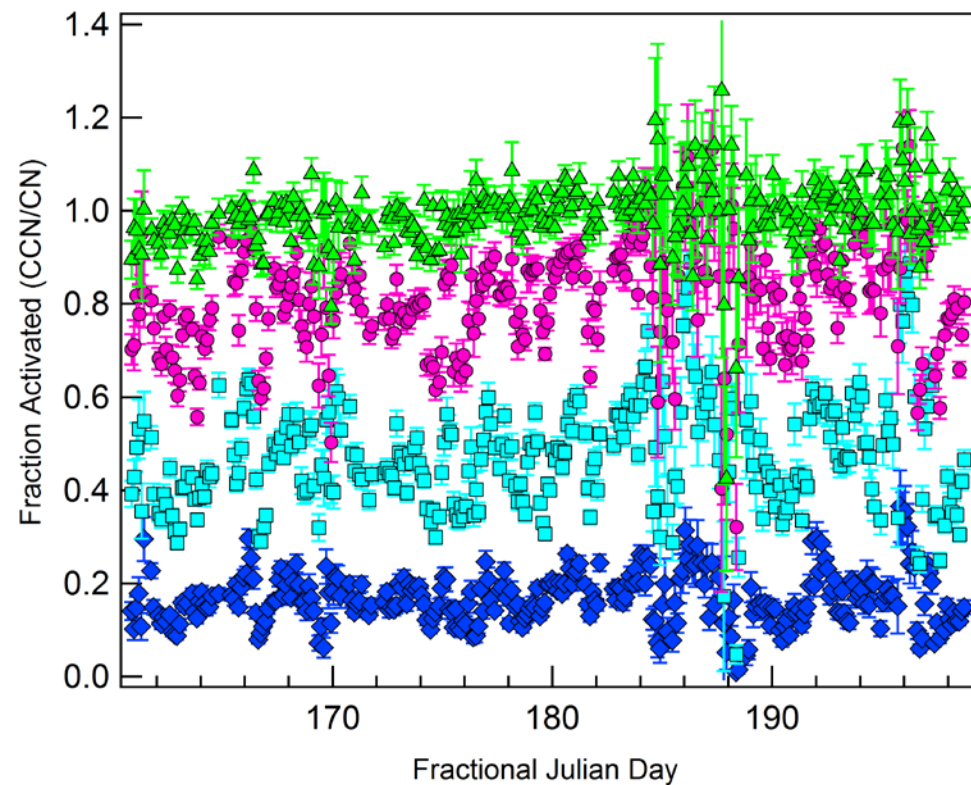
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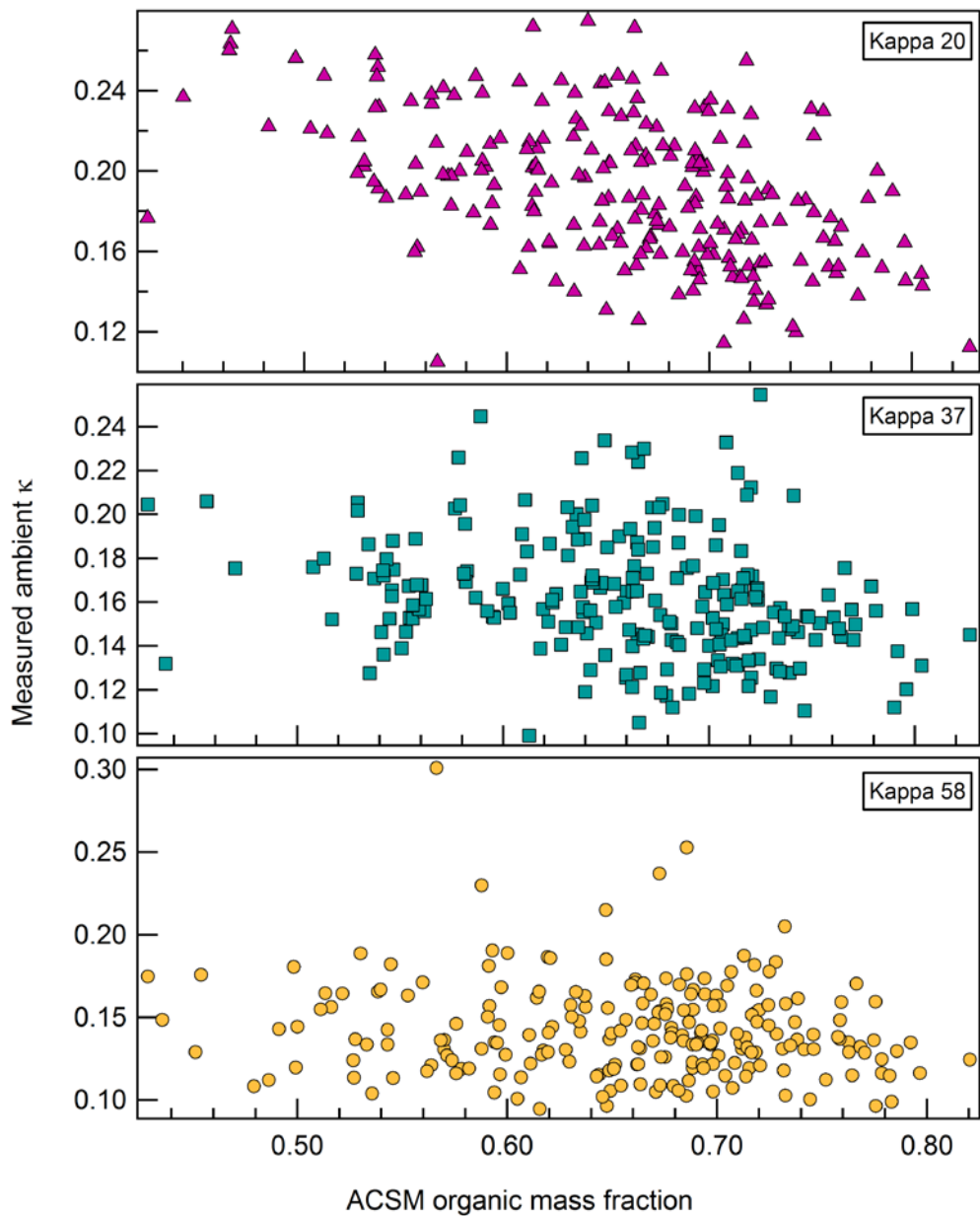
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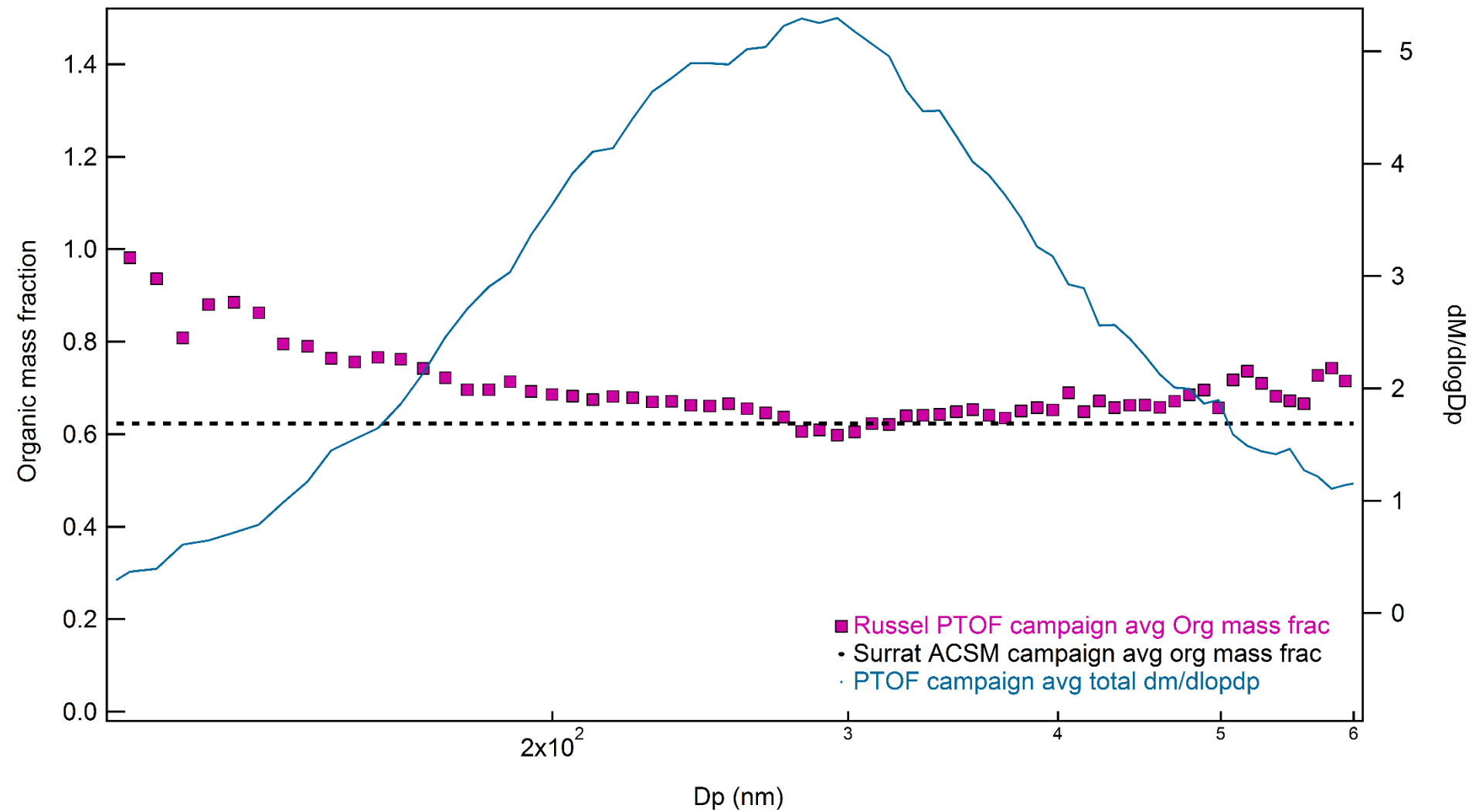
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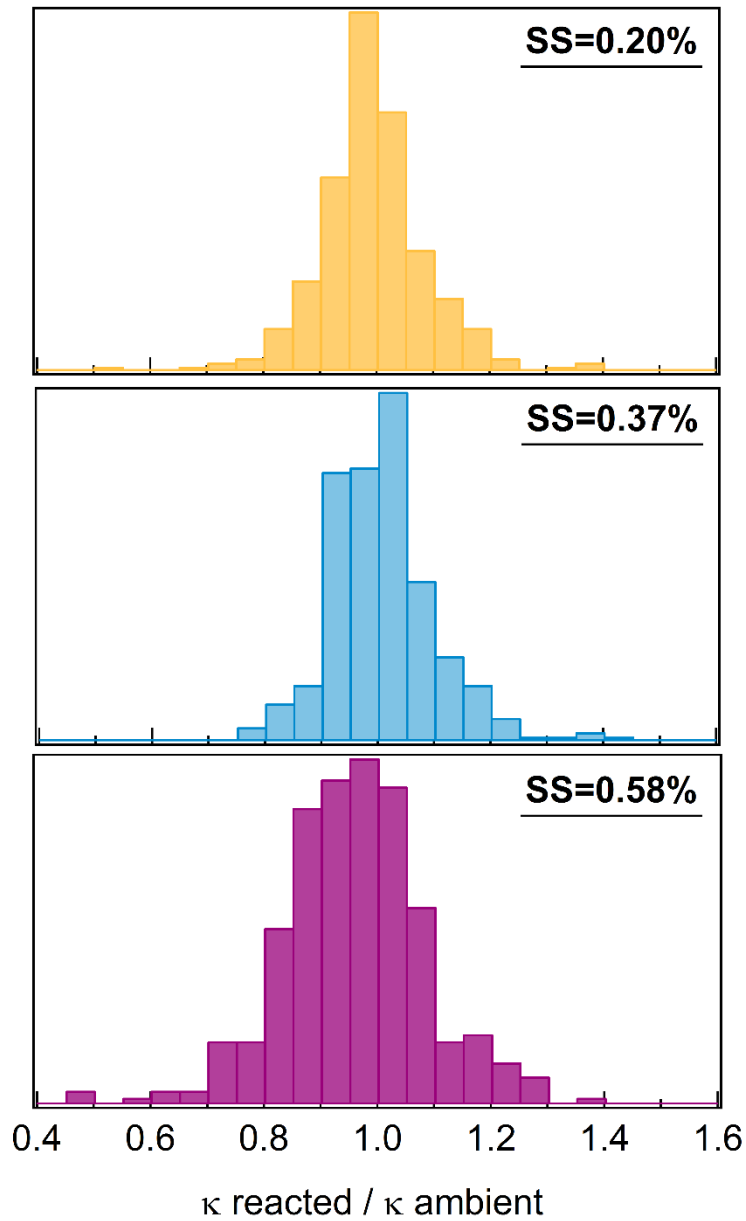
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